

# Ground Water Quality Assessment of Daur Taluka, Shaheed Benazir Abad

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**Abstract**—The aim of this study was to assess the ground water quality of Daur Taluka of district Shaheed Benazir Abad for drinking purposes. Forty groundwater samples were collected from different locations and brought to Pakistan Council of Research in Water Resources (PCRWR) for analyzing various groundwater physical, chemical and biological parameters. The results of this study revealed that color, pH, magnesium, alkalinity and nitrate of all samples were found within the permissible limits of World Health Organization (WHO). Analytical results revealed that the percentage of samples that were beyond WHO standards was 15% regarding taste, 42.5% regarding TDS, 20% regarding chlorine, 12.5% regarding sulfate 12.5 regarding sodium and 32.5% regarding hardness. Microbiological contamination was found positive in the 25% of samples. The findings of this study revealed that the most (82.5%) of the samples of the study area belong to the category of hard to very hard water and their nature were alkaline.

**Keywords**—water quality parameters; arsenic; Taluka Daur; Shaheed Benazir Abad; WHO standards

## I. INTRODUCTION

Water is one of the most important ecosystem and climate components. Water quantity and quality determine the biodiversity of an ecosystem. Water quantity and quality is not distributed evenly in the world [1, 2]. Safe drinking water is not only a fundamental requirement of all living organisms but also an indispensable human right [3, 4]. Insufficient availability of fresh water coerces people to use ground water for meeting their needs. About 70% of Pakistan households use ground water for domestic purposes [5, 6]. Water also becomes a vital source of increasing demand for agriculture and industry [7]. The quality of ground water in Sindh region is worsening due to overuse of fresh water layer. Sindh is an arid region, so

ground water is recharged only during flooding time along the water channel sides [8, 9]. About 78% of Sindh province has saline ground water whose salinity varies and at maximum is 9dS/m [7, 10]. The groundwater is polluted due to various natural and anthropogenic factors. The disposal of untreated effluents from industries and household sewage, seepage of fertilizers and land degradation are the main sources of groundwater pollution [11-14]. The quality of ground water of Sindh urban and suburban areas is very poor and chemically and bacteriologically not fit for drinking purposes [15]. The consumption of contaminated water causes various diseases [16]. About 98% of the households of Shaheed Benazir Abad are using ground water [17]. The aim of this study is to assess the ground water quality of four sites of Taluka Daur of Shaheed Benazir Abad for drinking purposes.

## II. METHODOLOGY

Forty groundwater samples were collected from different locations of four union councils of Taluka Daur (Daur, Bandhi, Jam Sahib and Bucheri) of district Shaheed Benazir Abad. Ten water samples have been taken from each union council. Sterilized plastic bottles were used for collecting the samples which were washed with distilled water twice prior to sampling. The samples were collected by using standard methods suggested in [18, 19]. The collected water samples were brought to Pakistan Council of Research in Water Resources (PCRWR) regional laboratory for analyzing regarding the following selected physical and chemical groundwater parameters (methods of analysis and concerned instruments employed for each water quality parameter are summarized in Table I).

### A. Physical Water Parameters under Study

These parameters of water are turbidity, electrical conductivity, pH, taste and color.

### B. Chemical Water Parameters under Study

These parameters were bicarbonate ( $\text{HCO}_3$ ), chlorine (Cl), sulfate ( $\text{SO}_4$ ), hardness, calcium (Ca), magnesium (Mg), sodium (Na), potassium (K), alkalinity ( $\text{CaCO}_3$ ), total dissolved solids (TDS) and nitrate ( $\text{NO}_3$ ).

### C. Biological Water Aspects under Study

Presence-absence test kit was used to evaluate the water samples for microbiological contamination. American Public Health Association's guidelines were followed for examining the water samples and calibrating the equipment and instruments [20].

### B. Taste and Odor

The taste and odor of 34 samples are non-objectionable. Only 6 samples possessed objectionable taste.

### C. TDS

TDS level is a good indicator of taste. Less than 300mg/l is excellent, in the range between 300 and 600mg/l is good, from 600 to 900mg/l is fair, between 900 and 1200mg/l is poor and greater than 1200mg/l is not acceptable [19]. The upper limit of TDS of drinking water should be 1000mg/l according to WHO standards. Figure 1 shows the results of TDS for all samples. Results indicate that the TDS of 23 (57.5%) samples is within the permissible limits. The range of TDS of the entire samples was 100-3290mg/l and the average value was 1273mg/l. Further analysis showed that 3 samples were excellent, 7 were good, 5 were fair and 7 were poor in taste while the remaining samples were not acceptable.

TABLE I. WATER QUALITY PARAMETERS AND METHODS OF ANALYSIS

S. No.	Water Quality Parameter	Method of analysis
1	TDS	2540C, Standard method (1992)
2	Conductivity ( $\mu\text{S}/\text{cm}$ )	E.C. meter, Syber Scan CON 11 Singapore
3	PH	Jenco Handheld pH meter, Model 6230N
4	Color (TCU)	Sensory Test
5	Taste	Sensory Test
6	Bicarbonate (mg/l)	2320, Standard method (1992)
7	Chloride(mg/l)	Titration Standard method (1992)
8	Sulfate	Spectrophotometer Optizen 2120UV plus Korea
9	Hardness (mg/l)	EDTA Titration Standard method (1992)
10	Calcium (mg/l)	3500-Ca-D, Standard method (1992)
11	Magnesium (mg/l)	234-C, Standard method (1992)
12	Sodium (mg/l)	Flame photometer Italy
13	Potassium	Flame photometer
14	Alkalinity	2320, Standard method (1992)
15	Turbidity	Turbidity meter Lovibond PC check kit Germany
16	Nitrate	Colorimeter, Hach- DR2800, USA
17	B.C	Presence /absence test kit

Statistical methods were used to find the minimum, maximum, mean and range of all water quality parameters. The results were compared and evaluated with the help of WHO guidelines for physical and chemical water characteristics.

## III. RESULTS AND DISCUSSION

### A. Color

Pure water is usually considered as colorless but it actually possesses a slightly blue hue. The sources of color in ground water are natural and anthropogenic. The particular geological formation or some metallic species of the region or seepage of effluent impact the groundwater color. WHO set the 15 TCU limit for the color of drinking water. All samples were aesthetically good and clear.

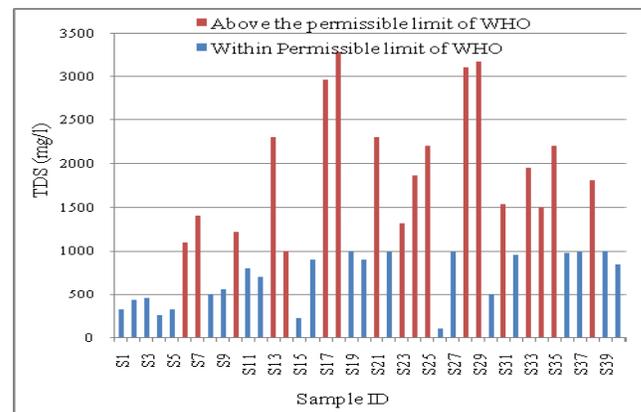


Fig. 1. Comparison of results of TDS of the entire samples

### D. Electrical Conductivity (EC)

No guide line value has been given by WHO and USEPA for EC. The range of EC of all 40 samples was found as 389-5140 $\mu\text{S}/\text{cm}$  and the average value was 2103.7 $\mu\text{S}/\text{cm}$ .

### E. pH

pH is a very important water quality parameter which describes its acidic or alkaline nature. Drinking water pH should range from 6.5 to 8.5 according to WHO standards. Results show that the pH of all samples is within permissible limits. The range of pH of all samples was 6.7-7.6 and the average value was 7.2. The pH values of 8 samples were below 7 and the values of the remaining samples were greater than 7. This shows that the nature of most of the ground water in the study area is alkaline.

### F. Bicarbonate

Bicarbonate is the main anion in ground water. It derives from  $\text{CO}_2$  released as a consequence of organic decay in the soil. No guide line value of bicarbonate ( $\text{HCO}_3$ ) has been given by WHO. The range of  $\text{HCO}_3$  of all tested 40 samples was 60-680mg/l and the average value was 295.25mg/l.

G. Chlorine (Cl)

There are the various Cl sources in groundwater. These sources are leaching from mineral rock or soil, seepage of domestic and industrial effluent, sea water intrusion etc. [21]. Ground water shows higher Cl concentration compared to surface water. Cl is essential for metabolic activity and other physiological processes of human body but excess concentration of Cl is also detrimental [5]. The highest desirable limit of chloride in drinking water is accredited as 250mg/l by WHO. Results of Cl concentration for all study area samples are shown in Figure 2

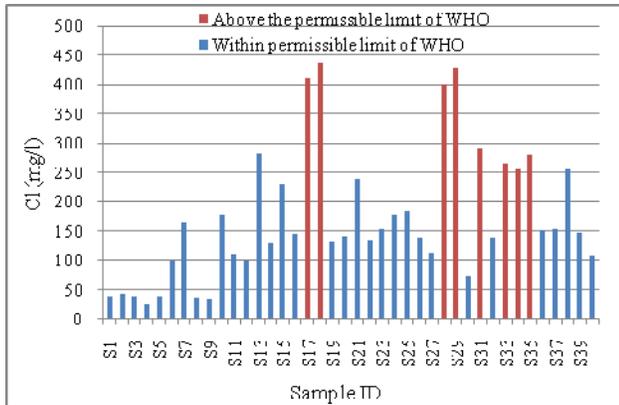


Fig. 2. Comparison of results of Cl of all samples

Figure 2 indicates that the Cl concentration of 32 samples is within the limit given by WHO while 20% of the samples are exceeding it. Water containing Cl in concentration greater than 250mg/l has salty taste. The range of Cl of all the samples was 25-437mg/l and the average value was 172.47mg/l.

H. Sulfate (SO<sub>4</sub>)

Laxative effect is associated with water ingestion containing elevated sulfate level [22]. The highest desirable limit of sulfate is acknowledged as 250mg/l in drinking water by WHO. Figure 3 shows the results of Sulfate (SO<sub>4</sub>) concentration in the study area samples.

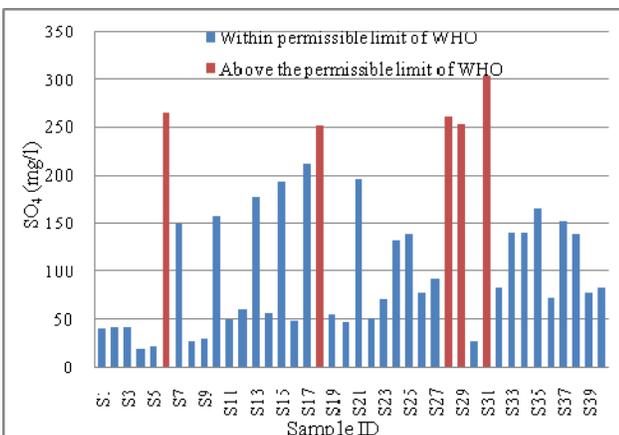


Fig. 3. Comparison of results of SO<sub>4</sub> of all 40 samples

Results exhibit that the Sulfate (SO<sub>4</sub>) concentration of 35 samples is within the maximum permissible limit specified by WHO. The sulfate concentration of 12.5% of the samples is above the permissible limit while the sulfate concentration of 10 samples was lower than the desirable limit of 50mg/l. The range SO<sub>4</sub> of the samples was 20-302mg/l and the average value was 115mg/l.

I. Magnesium (Mg)

Mg is one of the important minerals of water and is very essential for human health. The content of Mg in water is used to determine the quality of water for various purposes [5]. WHO has specified 150mg/l of Mg for drinking water. Mg concentrations of all samples are within the maximum permissible limit. The magnesium content varied from 10 to 121mg/l and the average value was 50.65mg/l.

J. Calcium (Ca)

Calcium is one of the most significant minerals of water which is indispensable for bones, teeth and cell physiology. Excess calcium consumption is very detrimental for human health [5]. WHO has not set any guide line value of Ca for drinking water. The range of Ca of all samples was 16-160mg/l and the average value was 71.5mg/l.

K. Hardness

Water hardness is a very important parameter determining its potential use. Various minerals are dissolved in water impacting its hardness. Water hardness is often calculated as calcium carbonate concentration. It should be 500mg/l in drinking water according to WHO. Drinking water has been classified on the basis of hardness as follows: Soft :between 0 and 75mg/l, moderately hard: between 75 and 150mg/l, hard: between 150 and 300mg/l and very hard: greater than 300mg/l [21]. The results of degree of hardness are shown in Figure 4.

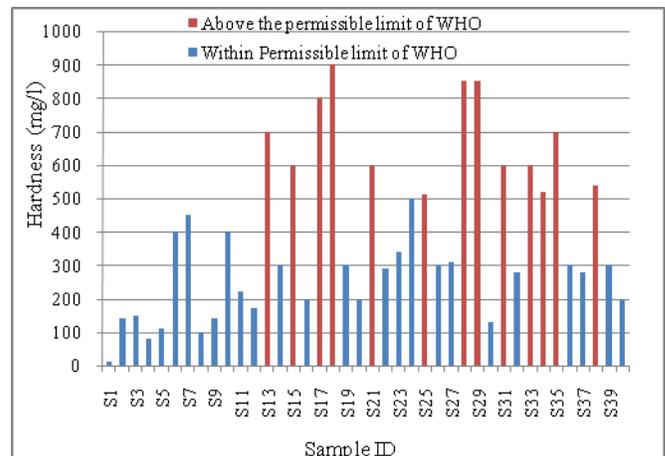


Fig. 4. Comparison of results of hardness of all 40 samples

Results indicate that the hardness of 27 samples is within the permissible limit, while 32.5% of the samples have hardness above the permissible WHO limit. The range of hardness of all 40 tested samples was 11-900mg/l and the

average value was 384.27mg/l. Only one groundwater sample was soft according to the above criteria. While 6 samples were moderately hard, 14 samples belonged to hard and the remaining 19 samples comprised very hard water. The analytical result shows that 82.5% of ground water samples of the study area fall in the category of hard to very hard water. Elevated level of water hardness creates heart and kidney problems [23].

#### L. Sodium (Na)

Sodium is one of the most common water metallic elements. Proper quantity of Na is very essential for the human body. Elevated Na level imparts its flavor to water and can be harmful for human health [24]. Its concentration should be 200mg/l in drinking water according to WHO. The results of Na concentration are shown in Figure 5.

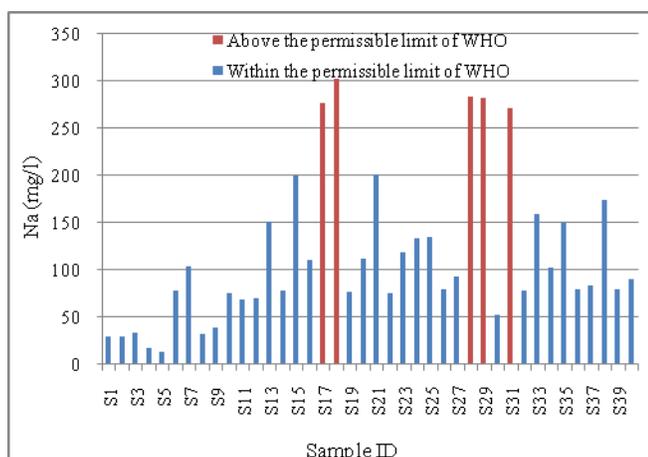


Fig. 5. Comparison of results of Na of all samples

Results indicate that the sodium concentration of 35 samples was found within the permissible limit given by WHO. The remaining 12.5% of the samples possessed higher Na concentration. The range of Na concentration of all samples was 14-301mg/l and the average value was 115.32 mg/l.

#### M. Potassium (K)

Potassium is an alkali element. It is very essential for all living organisms, it is particularly found in cell tissues and helps in hydration [25]. WHO has not specified a permissible value of K. However European Commission set 12mg/l in water for drinking purpose. The results of K of all samples are shown in Figure 6. Results indicate that K concentrations of 39 samples are within the permissible limit given by EC and only 1 sample is beyond it. The range of K of all samples was 1-13mg/l and the average value was as 4.61mg/l.

#### N. Alkalinity

Alkalinity is the presence of different components in water which make it alkaline. Alkalinity elevated level may create various health problems. No guide line value of alkalinity has been given by WHO. The range of alkalinity of the samples

was 1.2-13.6m.mol/l and the average value was calculated as 5.96m.mol/l.

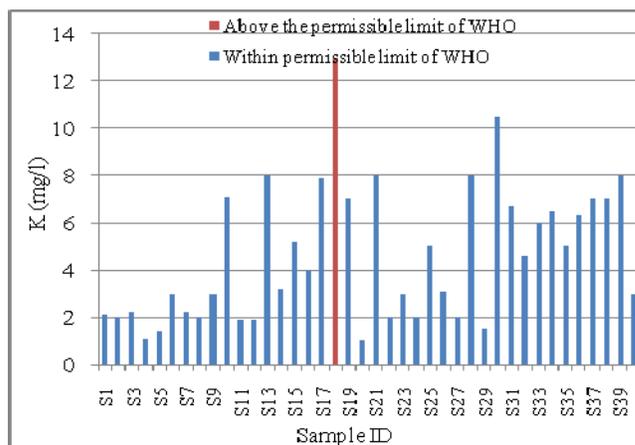


Fig. 6. Comparison of results of potassium of all samples

#### O. Nitrate (NO<sub>3</sub>)

Nitrate is an inorganic and water soluble compound derived from nitrogen and essential for plant life. The natural source of nitrate in groundwater is the atmosphere through the nitrogen cycle. Anthropogenic sources are leaching from industrial effluent, agricultural chemicals and sewage [26]. The consumption of groundwater having elevated level of nitrate may cause diabetes, thyroid disease, gastric cancer and methemoglobinemia [27]. WHO set the maximum permissible limit of nitrate as 10mg/l in drinking water. The results show that nitrate levels of all samples are within this limit. The range of nitrate of the samples was 0.25-2.7mg/l and the average value was 1.27mg/l.

#### P. Microbiological Contamination

The microbial contamination of groundwater is a severe cause of health concern. Dysentery, diarrhea, cholera, typhoid and polio are diseases which can be transmitted through contaminated water. The main sources of microbial contamination of water are human or animal feces and agricultural activity [27, 28]. Presence-absence test kit was used to evaluate the ground water samples for microbiological contamination. The results indicate that 75% of samples were free from microbiological contamination while 25% of samples showed positive results.

#### IV. CONCLUSIONS

Results of this study revealed that 82.5% of groundwater samples of the study area belong to the category of hard to very hard water and the most (82.5%) of the samples are alkaline in nature. Color, pH, magnesium, alkalinity and nitrate of all the samples are found within the permissible limits of WHO. The analytical results revealed that the percentage of samples beyond permissible limits, regarding the tested parameters were taste (15%), TDS (42.5%), Cl (20%), sulfate (12.5%), Na (12.5%) and hardness (32.5%). The microbiological contamination of 25% of the samples was found positive.

Groundwater with parameters beyond permissible limits should not be used for drinking purpose.

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