

# An Experimental Study on the Carbonation Resistance of Conventional Concretes in Vietnam

**Quynh Nga Le**

Structural Engineering Department, University of Transport and Communications, Hanoi, Vietnam  
ngaketcau@utc.edu.vn (corresponding author)

**Van Dinh Dao**

Structural Engineering Department, University of Transport and Communications, Hanoi, Vietnam  
daovandinhkc@utc.edu.vn

**Viet Hung Tran**

Structural Engineering Department, University of Transport and Communications, Hanoi, Vietnam  
hungtv@utc.edu.vn

Received: 20 March 2026 | Revised: 14 April 2026 and 4 May 2026 | Accepted: 15 May 2026

Licensed under a CC-BY 4.0 license | Copyright (c) by the authors | DOI: <https://doi.org/10.48084/etasr.18838>

## ABSTRACT

**Reinforced Concrete (RC) is widely used in global infrastructure development. Long-term performance of the RC structures is a primary objective for managers, engineers, and scientists. Global climate change leads to a significant rise in atmospheric CO<sub>2</sub> concentrations, particularly in developing nations like Vietnam, due to rapid urbanization. This study examines the carbonation resistance of concrete mixtures with water-to-cement ratios ranging from 0.35 to 0.55 and was conducted in a laboratory setting. Accelerated carbonation tests were performed at the Center for Cement and Concrete (Hanoi) in accordance with the latest standard. The results suggest that concrete with a Water-to-Cement ratio (W/C) greater than 0.50 has low carbonation resistance and should be used with caution in structural applications. Beyond this threshold, the carbonation depth increases rapidly.**

*Keywords-carbonation resistance; experimentation; CO<sub>2</sub>; penetration; corrosion*

## I. INTRODUCTION

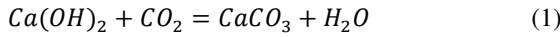
The rapid expansion of infrastructure in Vietnam has led to strict requirements for the durability and lifespan of Reinforced Concrete (RC) structures. Among the various factors, carbonation was identified as the primary cause of structural degradation, especially in urban and industrial areas. As the impact of global climate change escalates, atmospheric CO<sub>2</sub> concentrations — the most crucial factor influencing the carbonation process — have risen significantly. Authors in [1-3] estimated the time at which steel corrosion begins, as concrete's pH decreases from 13 to approximately 9. Authors in [4] indicated that pretreating materials with carbonation can significantly improve their mechanical properties in concrete mixtures, developing environmentally friendly concrete solutions [4]. Authors in [1] evaluated and compared predictive models for concrete carbonation depth, a key factor affecting the durability of reinforced concrete. Authors in [2] provided a thorough review of carbonation in cementitious materials, from fundamental mechanisms to contemporary analytical techniques, including Scanning Electron Microscopy (SEM).

Authors in [5] emphasized the need to search for rapid durability prediction methods as an alternative to traditional testing, enabling engineers to quickly compare the resistance of various concrete mixes and eliminating the need for month-long experimental durations. The hot and humid climate of Vietnam, with high average annual temperatures, accelerates carbonation rates compared to temperate regions. Consequently, empirical research on the carbonation resistance of common concrete types in Vietnam is significant to provide appropriate technical recommendations for design and construction. Such data also serve as a crucial input parameter for service life prediction models for engineering works.

## II. THEORETICAL BACKGROUND

Carbonation is the process by which atmospheric Carbon Dioxide (CO<sub>2</sub>) penetrates the pore network of concrete, dissolves in the pore solution to form weak carbonic acid (H<sub>2</sub>CO<sub>3</sub>), and reacts with cement hydration products. The primary reactions include:

- Reaction with calcium hydroxide  $\text{Ca}(\text{OH})_2$ : This reaction is important because it reduces the alkalinity of concrete:



After the  $\text{Ca}(\text{OH})_2$  content is depleted, the  $\text{CO}_2$  continues to attack the C-S-H gel.

- Reaction with Calcium Silicate Hydrate (C-S-H) gel: This process involves decalcifying the C-S-H phase, which changes the internal microstructure. According to Fick's first law of diffusion, the carbonation depth is a function of time. Using this law, the depth of carbonation in concrete is determined as:

$$x_c = \sqrt{\frac{2D_{\text{CO}_2}}{a} \Delta C_{\text{CO}_2} \sqrt{t}} \quad (2)$$

where  $x_c$  is the carbonation depth,  $\Delta C_{\text{CO}_2}$  is the carbon dioxide concentration difference, and:

$$a = 0.75 C_{\text{CaO}} C_c \alpha_h \frac{M_{\text{CO}_2}}{M_{\text{CaO}}} \quad (3)$$

where  $a$  is the binding capacity ( $\text{kg}/\text{m}^3$ ) of concrete [6, 7],  $C_{\text{CaO}}$  is the Calcium oxide content in the cement,  $C_c$  is the cement content (mass of cement) in the concrete, and  $\alpha_h$  is the degree of hydration [8]:

$$\alpha_h(t) = \frac{t}{2.0+t} \left[ \frac{1.031W/C}{0.194+W/C} \right] \quad (4)$$

where  $M_{\text{CO}_2}$  is the molecular weight of carbon dioxide, and  $M_{\text{CaO}}$  is the molecular weight of calcium oxide. The carbonation rate of concrete depends on several factors, including the Water-to-Cement ratio (W/C) [9, 10], ambient  $\text{CO}_2$  concentration [6], porosity [8], humidity, ambient temperature [9, 11], and the concrete's compressive strength [12].

### III. CARBONATION DEPTH ANALYSIS

#### A. Diffusion Coefficient

Authors in [13] proposed a formula for calculating carbonation depth, presenting the effective  $\text{CO}_2$  diffusion coefficient in carbonated concrete as:

$$D_{\text{CO}_2} = 6.1 \times 10^{-6} \left( \frac{\varepsilon_c}{\frac{W_c}{\rho_c} + \frac{W_p}{\rho_p} + \frac{W_w}{\rho_w}} \right)^3 \left( 1 - \frac{H}{100} \right)^{2.2} \quad (5)$$

where  $\varepsilon_c$  is the porosity of concrete,  $W_c$ ,  $W_p$ ,  $W_w$  are the contents of cement, supplementary materials, and water,  $\rho_c$ ,  $\rho_p$ ,  $\rho_w$  are the densities of cement, additives, and water, respectively, and  $H$  is the ambient humidity. The  $\text{CO}_2$  binding capacity  $a$  is determined as:

$$a = 0.33 C_{\text{CH}} + 0.214 C_{\text{CSH}} \quad (6)$$

where  $C_{\text{CSH}}$  is the C-S-H content, and  $C_{\text{CH}}$  is the  $\text{Ca}(\text{OH})_2$ .

#### B. Vietnamese Standards

According to the Vietnamese standard [14], the evolution of carbonation depth over time in experimental research is typically expressed as:

$$x_c = k\sqrt{t} \quad (7)$$

where  $x_c$  is the carbonation depth (mm),  $k$  is the carbonation rate coefficient ( $\text{mm}/\sqrt{\text{week}}$ ), which reflects the material's resistance to  $\text{CO}_2$  ingress, and  $t$  is the exposure time (weeks).

#### C. Model Code for Service Life Design

According to [9], the carbonation depth is calculated as:

$$x_c(t) = \sqrt{2k_e k_c (k_t R_{\text{ACC},0}^{-1} + \varepsilon_t) C_s \sqrt{t} W(t)} \quad (8)$$

where  $x_c(t)$  is the carbonation depth at time  $t$  (mm), and  $k_c$  is the environmental function depending on the ambient humidity, /.

$$k_e = \left[ \frac{1 - (\text{RH}_{\text{real}})^5}{1 - (\text{RH}_{\text{ref}})^5} \right]^{2.5} \quad (9)$$

$\text{RH}_{\text{real}}$  is the relative humidity of the carbonated layer (%), and  $\text{RH}_{\text{ref}}$  is the reference relative humidity used in the concrete carbonation resistance test ( $\text{RH}_{\text{ref}}=65\%$ ).

$$k_c = \left( \frac{t_c}{7} \right)^{b_c} \quad (10)$$

where  $t_c$  is the period of curing (day),  $b_c$  is the exponent of regression ( $b_c$  follows a normal distribution with a mean value of -0.567 and a standard deviation  $s = 0.024$ ),  $k_t$  is a regression parameter (with a mean value of 1.25 and a standard deviation  $s = 0.35$ ), and  $R_{\text{ACC}}^{-1}$  is the inverse effective carbonation resistance of the concrete [ $(\text{mm}^2/\text{s})/(\text{kg}/\text{m}^3)$ ]:

$$R_{\text{ACC}}^{-1} = \left( \frac{x_c}{\tau} \right)^2 \quad (11)$$

where  $x_c$  is the carbonation depth (m), and  $\tau$  is the time constant [ $(\text{s}/\text{kg}/\text{m}^3)$ ]<sup>0.5</sup> depending on the specific test conditions:

$$\tau = \sqrt{2\Delta C_{\text{CO}_2} t} \quad (12)$$

For an accelerated test with 2%  $\text{CO}_2$  concentration over 28 days  $\tau = 420$  [ $(\text{s}/\text{kg}/\text{m}^3)$ ]<sup>0.5</sup>,  $\varepsilon_t$  is the error term [ $(\text{mm}^2/\text{years})/(\text{kg}/\text{m}^3)$ ] with a mean value of 315.5 and a standard deviation  $s = 48$ ,  $C_s$  is the  $\text{CO}_2$  concentration ( $\text{kg}/\text{m}^3$ ), and  $W(t)$  is the weather function.

#### D. Carbonation Depth Model

Authors in [15] found that the carbonation depth model is:

$$x_c = k\sqrt{t} = n_1 n_2 n_3 n_4 n_5 k_{av} \sqrt{t} \quad (13)$$

where  $x_c$  is the depth of the carbonated concrete layer (mm),  $n_1$  is the carbonation front condition parameter ( $n_1=1.9$  for all cases),  $n_2$  is the exposure condition parameter ( $n_2=0.6$  for outdoor exposure to rain;  $n_2=1.0$ ; for sheltered outdoor conditions, and  $n_2=1.0$  for indoor conditions).

#### E. Carbonation Model

Authors in [6] proposed the following model for calculating the carbonation depth in concrete:

$$x_c(t) = \sqrt{\frac{2D_{\text{CO}_2}(t)}{a} C_{\text{CO}_2}(t) t} \left( \frac{t_0}{t} \right)^{n_m} \quad (14)$$

$$D_{\text{CO}_2}(t) = D_1 t^{-n_d} \quad (15)$$

$$a = 0.75C_eCaO\alpha_H \frac{M_{CO_2}}{M_{CaO}} \quad (16)$$

where  $x_c$  is the carbonation depth (cm),  $t_0$  is the reference time (typically:  $t_0=1$  year),  $t$  is the time (years),  $C_{CO_2}(t)$  is the time-dependent mass concentration of ambient  $CO_2$  ( $10^{-3}$  kg/m<sup>3</sup>),  $D_{CO_2}(t)$  is the diffusion coefficient of  $CO_2$  in concrete,  $D_1$  is the  $CO_2$  gas diffusion coefficient after one year,  $n_d$  is the age exponent for the  $CO_2$  diffusion coefficient,  $C_e$  is the cement content (kg/m<sup>3</sup>), CaO is the Calcium oxide content in the cement,  $\alpha_H$  is the degree of hydration,  $M_{CaO}$  is the molar mass of CaO ( $M_{CaO} = 56$ ),  $M_{CO_2}$  is the molar mass of  $CO_2$  ( $M_{CO_2} = 44$ ), and  $n_m$  is an age factor of microclimate conditions, which is associated with the frequency of wetting-drying cycles ( $n_m=0$  for sheltered outdoor;  $n_m \leq 0.3$  for unsheltered outdoor).

F. Relationship between  $CO_2$  Diffusion Coefficient, Inverse Carbonation Resistance and Carbonation Depth

Substituting (2) into (8) and squaring both sides gives:

$$\frac{2D_{CO_2}\Delta CO_2 t}{a} = 2k_e k_c (k_t R_{ACC}^{-1} + \epsilon_t) C_s t W(t) \quad (17)$$

$$D_{CO_2} = a k_e k_c (k_t R_{ACC}^{-1} + \epsilon_t) W(t) \quad (18)$$

$$D_{CO_2} = a k_e k_c [k_t \left(\frac{x_c}{\tau}\right)^2 + \epsilon_t] W(t) \quad (19)$$

Under experimental conditions, the coefficients are considered as  $k_e=1, k_c=1, k_t=1, W(t)=1, \epsilon_t=0$ .

IV. EXPERIMENTAL STUDY TO DETERMINE THE CARBONATION RESISTANCE OF ORDINARY CONCRETE

A carbonation resistance test was conducted to evaluate concrete's ability to impede  $CO_2$  penetration and protect the reinforcement from corrosion caused by reduced pH in the pore solution. The primary objective is to determine the inverse carbonation resistance coefficient ( $R_{ACC}^{-1}$ ) in order to design the concrete cover thickness and predict the structure's service life according to the model code for service life design [9]. This provides an important scientific basis for ensuring the durability and safety of infrastructure amid increasing atmospheric  $CO_2$  concentrations.

A. Material Preparation

PC40 grade cement from the But Son cement plant was used. The coarse aggregate (crushed stone) was taken from Kien Khe in Ha Nam, whereas the fine aggregate, or yellow sand, came from the Lo River in Phu Tho. The superplasticizer was Sika Viscocrete 3000-20M. These materials were selected for the concrete mix proportions displayed in the Table I.

TABLE I. MIX PROPORTIONS (KG/M<sup>3</sup>) OF CONCRETE

Mix ID	Compressive strength (MPa)	Cement (kg)	Sand (kg)	Stone (kg)	Water (lt)	Admixture (lt)	W/C
C1	50	517	646	1066	181	5.7	0.35
C2	45	452.4	701	1066	181	5.0	0.40
C3	40	402.1	744	1066	181	4.0	0.45
C4	25	403.4	735	1017	202	3.2	0.50
C5	20	366.8	766	1017	202	2.9	0.55

Prismatic concrete test specimens with dimensions of 100 × 100 × 400 mm are illustrated in Figure 1 and Table II.

B. Experimental Procedure

Compressive strength testing was performed on the concrete specimens at 28 days after casting in compliance with ASTM C39/C39M-24 [16], as presented in Table III.



Fig. 1. Fabrication of specimens for carbonation resistance determination.

TABLE II. NUMBER OF TEST SPECIMENS FOR CARBONATION RESISTANCE DETERMINATION

Mix ID	W/C	Cylindrical	Prismatic
C1	0.35	3 (C11; C12; C13)	3 (CP11; CP12; CP13)
C2	0.4	3 (C21; C22; C23)	3 (CP21; CP22; CP23)
C3	0.45	3 (C31; C32; C33)	3 (CP31; CP32; CP33)
C4	0.5	3 (C41; C42; C43)	3 (CP41; CP42; CP43)
C5	0.55	3 (C51; C52; C53)	3 (CP51; CP52; CP53)

TABLE III. COMPRESSIVE STRENGTH TEST RESULTS FOR CYLINDRICAL SPECIMENS

Mix ID	Specimen ID	W/C	Individual compressive strength (MPa)	Average compressive strength (MPa)
C1	C11	0.35	63.0	64.0
	C12		67.6	
	C13		61.3	
C2	C21	0.4	55.6	56.4
	C22		60.9	
	C23		52.7	
C3	C31	0.45	50.9	51.3
	C32		51.8	
	C33		51.4	
C4	C41	0.5	39.9	40.0
	C42		37.9	
	C43		42.0	
C5	C51	0.55	27.9	29.0
	C52		28.9	
	C53		30.1	



Fig. 2. Accelerated carbonation chamber.

The prismatic specimens were coated and placed in an accelerated carbonation chamber set to 27°C, with a CO<sub>2</sub> concentration of 4% and a Relative Humidity (RH) of 65%, as shown in Figure 2.

Multiple measurements of carbonation depth at various time intervals, including 1, 4, 6, and 8 weeks, were conducted. At earlier stages, specifically for mixes with low W/C ratios (0.35 and 0.40), the carbonation depth was virtually immeasurable and the carbonation depths were not recorded for all mix proportions until the 8-week mark. After eight weeks, the specimens were split, and the carbonation depth was measured using a phenolphthalein indicator solution [17], as portrayed in Figure 3.

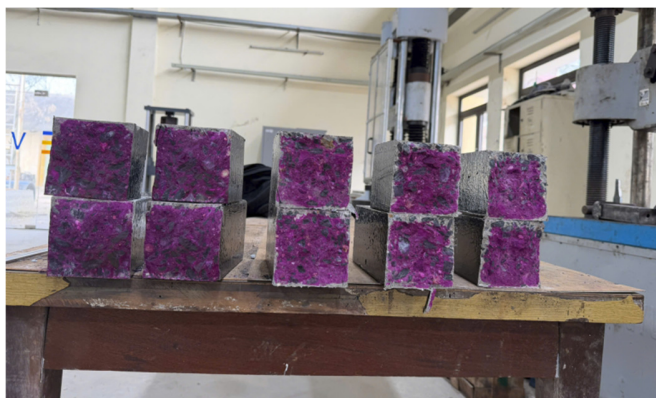


Fig. 3. Measurement of carbonation depth.

V. RESULTS AND DISCUSSION

The results obtained from the experiment are outlined in Table IV.

TABLE IV. ACCELERATED CARBONATION TEST RESULTS

Specimen ID	W/C	x <sub>c</sub> (mm)	R <sub>Acc</sub> <sup>-1</sup> (×10 <sup>-11</sup> ) ((m <sup>2</sup> /s)/(kg/m <sup>3</sup> ))	D <sub>CO<sub>2</sub></sub> (×10 <sup>-9</sup> ) (m <sup>2</sup> /s)	COV (%)
C1	0.35	1.552	0.3406	0.43193	12.56
C2	0.4	3.202	1.4496	1.68386	13.12
C3	0.45	4.916	3.4169	3.66058	11.74
C4	0.5	7.854	8.7214	9.66483	14.23
C5	0.55	11.858	19.8807	20.5545	12.89

A. Compressive Strength and Carbonation Depth of Concrete

Figure 4 illustrates the relationship between the compressive strength and carbonation depth of concrete. Based on this:

- The experimental data follow a linear regression model within the investigated range:
 
$$y = -0.2942x + 20.04 \tag{20}$$
- The coefficient of determination, R<sup>2</sup>, is exceptionally high at 0.9929, showing that compressive strength is an accurate predictor of the carbonation depth of the concrete samples under study.

- The experimental curve shows strong agreement with [12, 18], particularly in the higher strength range (>40 MPa), revealing that controlling compressive strength is essential to enhancing the carbonation resistance of concrete structures.

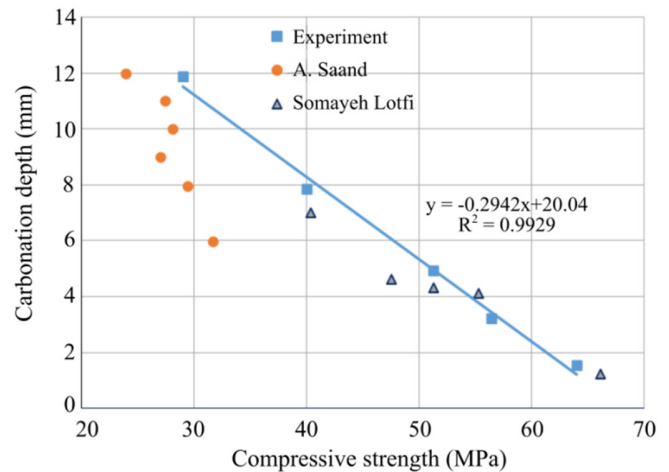


Fig. 4. Relationship between compressive strength and carbonation depth.

B. Water-to-Cement Ratio and Inverse Carbonation Resistance

Figure 5 shows the relationship between the water-to-cement ratio and inverse carbonation resistance. Based on this:

- The regression trendline for the experiment data follows an exponential power law:
 
$$y = 4.10^{-8}x^{8.8299} \tag{21}$$
- The coefficient of determination R<sup>2</sup> = 0,9998 indicates that this mathematical model is highly accurate and perfectly reflects the correlation within the experimental data. The rate of increase for R<sub>Acc</sub><sup>-1</sup> becomes significantly sharper once the w/c ratio exceeds 0.5.
- The performance characteristics of the concrete testing fall within the established ranges for the standard types of cement defined by the Fib Model Code [9], which validates the experimental methodology employed at the Vietnam Institute for Building Materials.
- To ensure the design service life of RC structures in Vietnam, W/C ratio should be less than 0.5.

C. Water-to-Cement Ratio and the CO<sub>2</sub> Diffusion Coefficient

Figure 6 demonstrates the relationship between the water-to-cement ratio and the CO<sub>2</sub> diffusion coefficient. Based on this:

- The experimental data follow a clear exponential law relationship:
 
$$y = 3.10^{-6}x^{8.4193} \tag{22}$$

- The coefficient of determination,  $R^2$ , is exceptionally high at 0.9991, indicating an agreement between the mathematical model and the experimental findings.
- An increase in the w/c ratio leads to a higher  $\text{CO}_2$  diffusion coefficient, thereby enhancing the susceptibility of concrete to carbonation-induced corrosion.

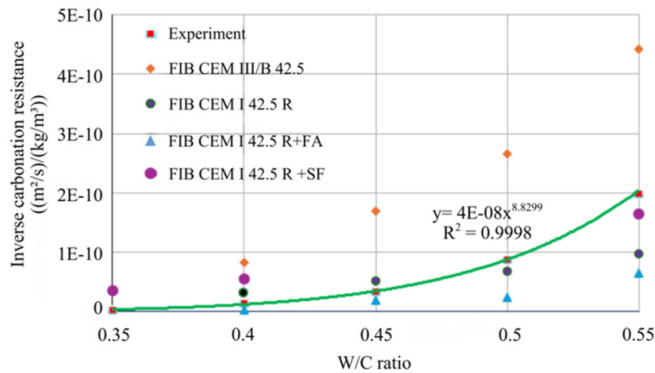


Fig. 5. Relationship between inverse carbonation resistance ( $R_{acc}^{-1}$ ) and W/C ratio.

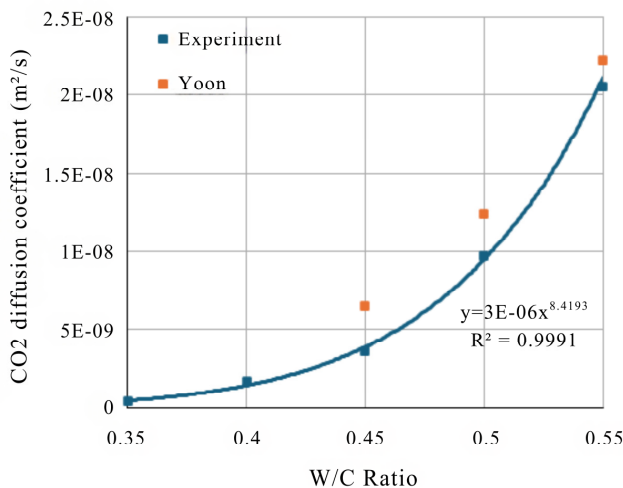


Fig. 6. Relationship between  $\text{CO}_2$  diffusion coefficient and W/C ratio.

## VI. CONCLUSIONS

This study provides a systematic overview of concrete carbonation resistance, emphasizing the importance of accelerated carbonation tests in determining input parameters for service life prediction models. The research results lead to the following conclusions:

- The experimental data indicate that the inverse carbonation resistance of the concrete ranges from  $3.4 \times 10^{-12}$  to  $19.88 \times 10^{-11}$  ( $\text{m}^2/\text{s}/(\text{kg}/\text{m}^3)$ ) and the diffusion coefficients range from  $4.32 \times 10^{-10}$  to  $2.055 \times 10^{-8}$  ( $\text{m}^2/\text{s}$ ), corresponding to Water-to-Cement ratios (W/C) between 0.35 and 0.55.
- The carbonation resistance strongly depends on the w/c ratio.

- These experimental data serve as a basis for researching the prediction of corrosion initiation time caused by carbonation in Reinforced Concrete (RC) structures in Vietnam.

## DECLARATION OF COMPETING INTERESTS

The authors declare no competing financial interests.

## ACKNOWLEDGMENT

This research is funded by the University of Transport and Communications (UTC) under grant number T2026-CT-027.

## DATA AVAILABILITY

The original laboratory data are presented within the paper. Data from [6, 9, 12] are plotted in Figures 4–6 strictly for comparison.

## REFERENCES

- [1] X. Wang, Q. Yang, X. Peng, and F. Qin, "A review of concrete carbonation depth evaluation models," *Coatings*, vol. 14, no. 4, 2024, Art. no. 386, <https://doi.org/10.3390/coatings14040386>.
- [2] Q. Qiu, "A state-of-the-art review on the carbonation process in cementitious materials: Fundamentals and characterization techniques," *Construction and Building Materials*, vol. 247, Jun. 2020, Art. no. 118503, <https://doi.org/10.1016/j.conbuildmat.2020.118503>.
- [3] A. Merah, "Methods of concrete accelerated carbonation test: A review," *Discover Civil Engineering*, vol. 1, no. 1, Aug. 2024, Art. no. 53, <https://doi.org/10.1007/s44290-024-00057-z>.
- [4] P. Chavan and C. Mishra, "Effect of carbonation on concrete structure: A review," *Journal of Physics: Conference Series*, vol. 3076, Aug. 2025, Art. no. 012008, <https://doi.org/10.1088/1742-6596/3076/1/012008>.
- [5] V. Shah, J. Mackechnie, and A. Scott, "Determination of carbonation resistance of concrete through a combination of cement content and tortuosity," *Journal of Building Engineering*, vol. 60, Nov. 2022, Art. no. 105176, <https://doi.org/10.1016/j.jobee.2022.105176>.
- [6] I.-S. Yoon, O. Çopuroglu, and K.-B. Park, "Effect of global climatic change on carbonation progress of concrete," *Atmospheric Environment*, vol. 41, pp. 7274–7285, 2007, <https://doi.org/10.1016/J.ATMOSENV.2007.05.028>.
- [7] S. Lay and P. Schießl, *LIFECON Deliverable D 3.2: Service Life Models*. Munich, Germany: Technische Universität München, Nov. 2003.
- [8] S. Cha, "Modeling of hydration process and analysis of thermal and hygral stresses in hardening concrete," Ph.D. dissertation, Seoul National University, Seoul, South Korea, 1999.
- [9] *Model Code for Service Life Design*, fib Bulletin 34. Lausanne, Switzerland: International Federation for Structural Concrete, 2006.
- [10] X. Wang, M. Nguyen, M. G. Stewart, M. Syme, and A. Leitch, Analysis of Climate Change Impacts on the Deterioration of Concrete Infrastructure – Part I: Mechanisms, Practices, Modelling and Simulations – A review. Canberra, Australia: CSIRO, 2010.
- [11] T. Du *et al.*, "Time-dependent model for in-situ concrete carbonation depth under combined effects of temperature and relative humidity," *Case Studies in Construction Materials*, vol. 22, Jul. 2025, Art. no. e04379, <https://doi.org/10.1016/j.cscm.2025.e04379>.
- [12] A. Saand, M. A. Keerio, and D. K. Bangwar, "Effect of Soorh Metakaolin on Concrete Compressive Strength and Durability," *Engineering, Technology & Applied Science Research*, vol. 7, no. 6, pp. 2210–2214, Dec. 2017, <https://doi.org/10.48084/etasr.1494>.
- [13] V. G. Papadakis, "Effect of supplementary cementing materials on concrete resistance against carbonation and chloride ingress," *Cement and Concrete Research*, vol. 30, no. 2, pp. 291–299, Feb. 2000, [https://doi.org/10.1016/S0008-8846\(99\)00249-5](https://doi.org/10.1016/S0008-8846(99)00249-5).

- [14] *TCVN 13929:2024 Concrete – Test Method for Accelerated Carbonation*,. Hanoi, Vietnam: Vietnam Institute for Building Materials, 2024.
- [15] M. G. Richardson, *Fundamentals of Durable Reinforced Concrete*. London, UK: CRC Press, 2023.
- [16] *C39/C39M-21 Standard Test Method for Compressive Strength of Cylindrical Concrete Specimens*. West Conshohocken, PA, USA: ASTM International, 2023.
- [17] *TCVN 13933:2024 Concrete – Test Method for Measuring Carbonation Depth*,. Hanoi, Vietnam: Vietnam Institute for Building Materials, 2024.
- [18] S. Lotfi, "C2CA concrete recycling process," Ph.D. dissertation, Delft University of Technology, Delft, The Netherlands, 2016.